

Solving reverse water-gas shift (RWGS) challenges with DRIFTS

The reverse water-gas shift (RWGS) reaction converts CO₂ and H₂ into CO and H₂O. It is crucial for syngas, methanol, and synthetic fuel production in petrochemicals, chemical manufacturing, and green fuels.



Growing demand for syngas, methanol and synthetic fuels

Global syngas demand

11.3%

CAGR: 11.3% (2024-2030)^[1]

Methanol demand

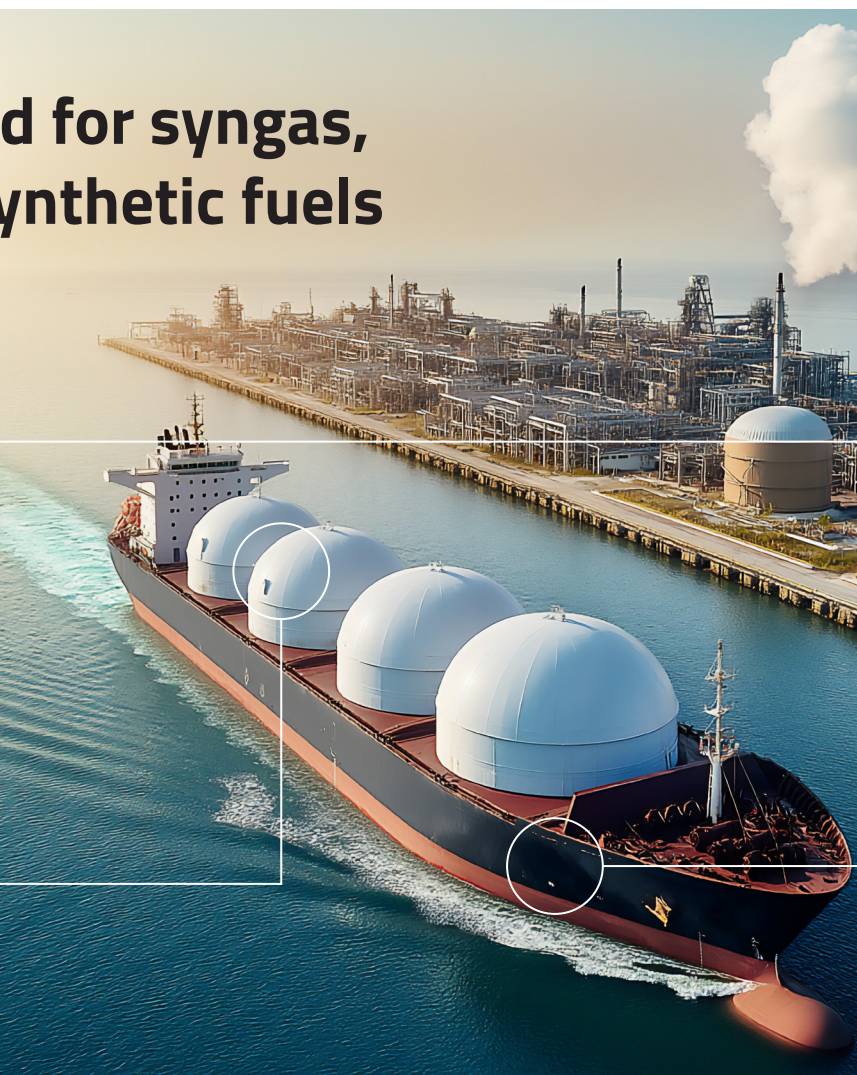
Production nearly doubled in the past decade:

98 Mt

million tonnes currently produced per annum

500 Mt

million tonnes by 2050^[2]



Synthetic fuel market

Market value:

\$48.4bn

in 2025

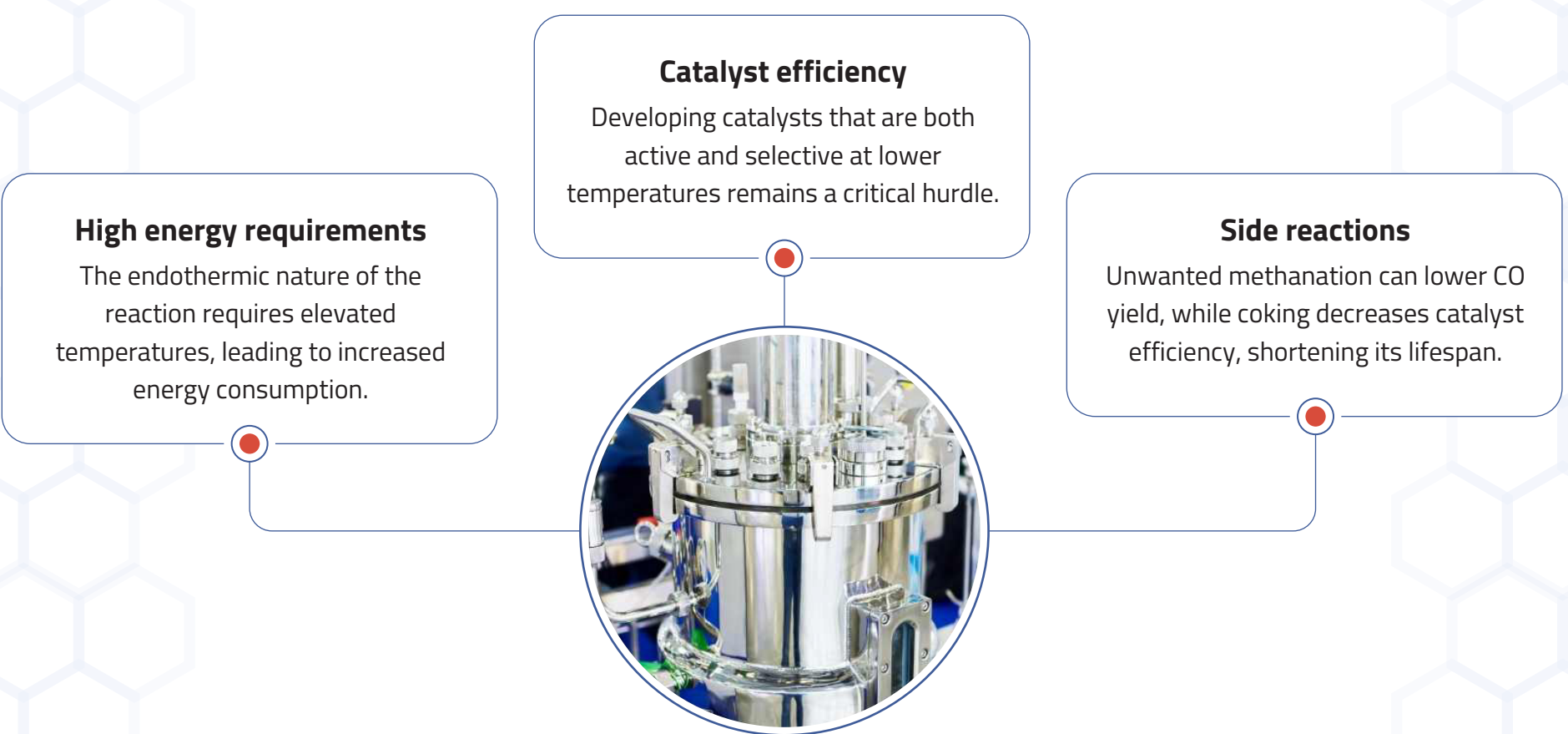
\$70.1bn

in 2030

(CAGR: 5.4%)^[3]

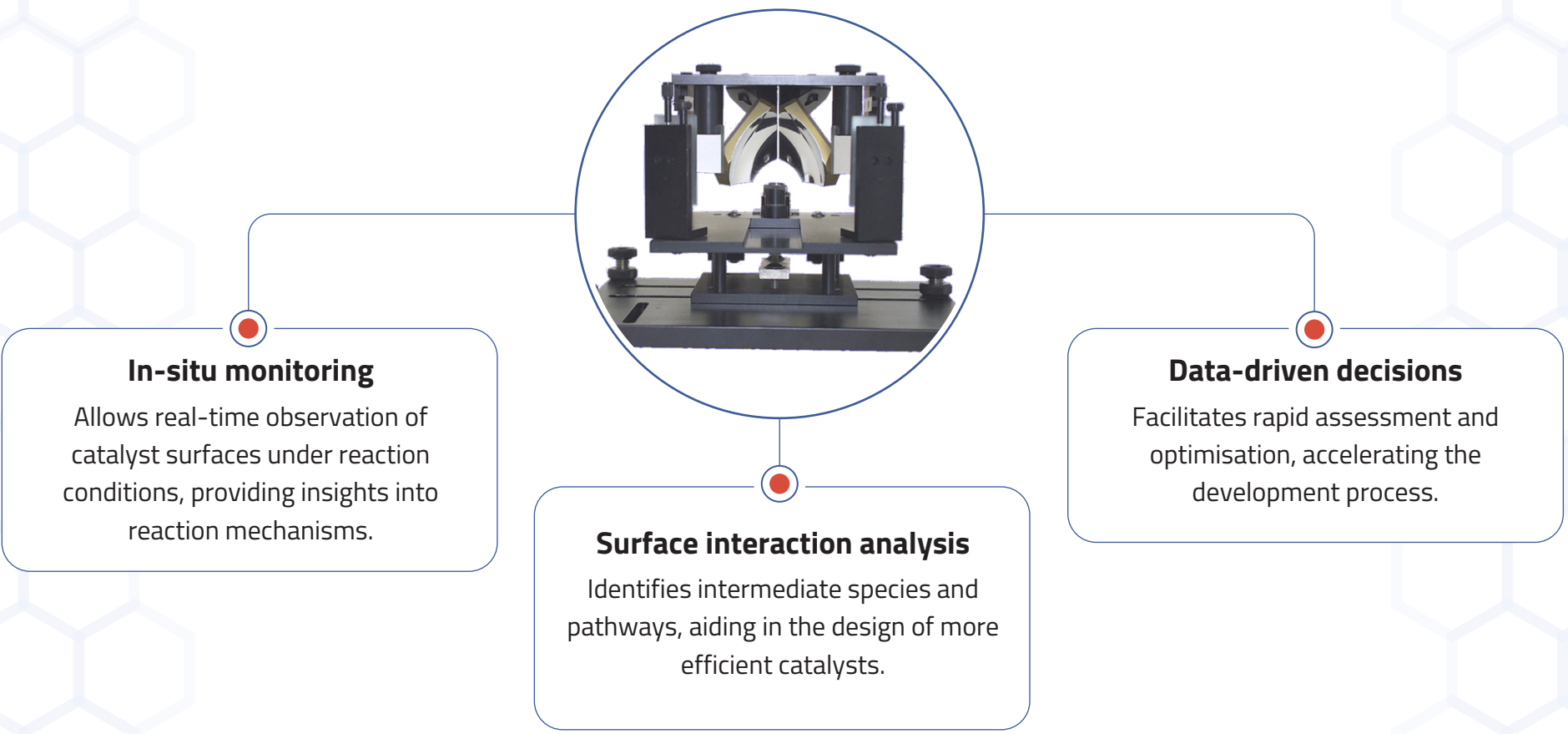
Challenges with the RWGS reaction

Despite its significance, the RWGS reaction faces some challenges.



Advancing RWGS research with DRIFTS

Diffuse reflectance infrared Fourier transform spectroscopy (DRIFTS) is instrumental in advancing RWGS research.



Innovating catalyst development through the RWGS

Recent research using DRIFTS has led to significant advancements and critical findings.

150°C

TEMPERATURES

Low-temperature catalysis
Iron-based catalysts have demonstrated high activity and selectivity for the RWGS reaction at temperatures as low as 423 K (150°C), reducing energy use.^[4]

Ti³⁺

FASTER RATES

Efficiency and faster reaction times
Gold-based catalysts boost RWGS efficiency. Au/TiO₂ catalysts convert CO₂ more effectively by using Ti³⁺ sites and oxygen vacancies. Operando UV-Vis confirms that more Ti³⁺ leads to faster reaction rates.^[7]

100

HOUR STABILITY

Enhanced catalyst stability
Nickel-based catalysts are showing superior performance; achieving near-theoretical CO₂ conversions at 450°C (44%) and 500°C (49%). 100-hour durability tests confirm long-term catalyst stability to help drive operation costs down.^[5]

x2.6

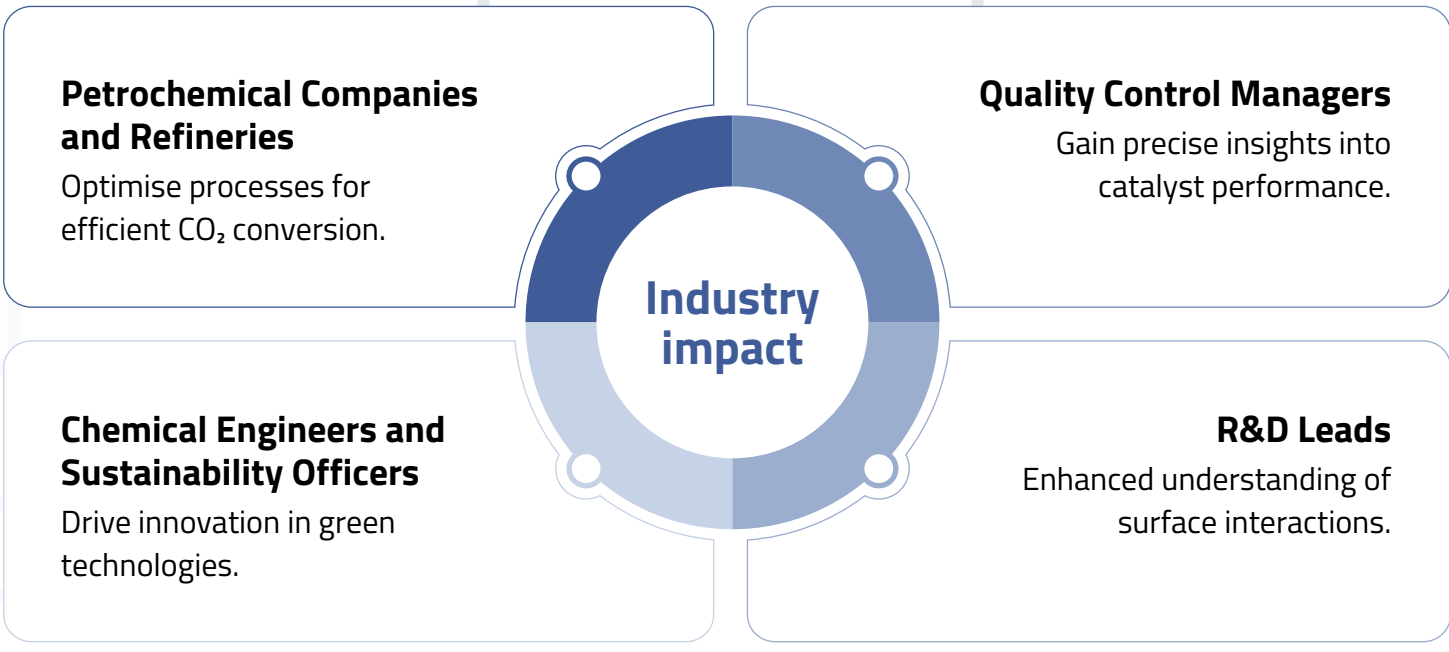
IMPROVEMENT

Improved CO₂ conversion
A 1.51 Pt:Co ratio boosted CO₂ conversion 2.6 times more than pure Pt at 500°C in RWGS reaction, with over 98% CO selectivity, reducing unwanted methane.^[6]

x2

LIFESPAN

Doubling catalyst lifespan
Adding a small amount of silicon to Cu/ZnO methanol catalysts slows down wear and tear, making them last twice as long. This boosts the RWGS reaction, making CO₂-to-methanol conversion more efficient and sustainable.^[8]



The Praying Mantis™: the performance standard in DRIFTS

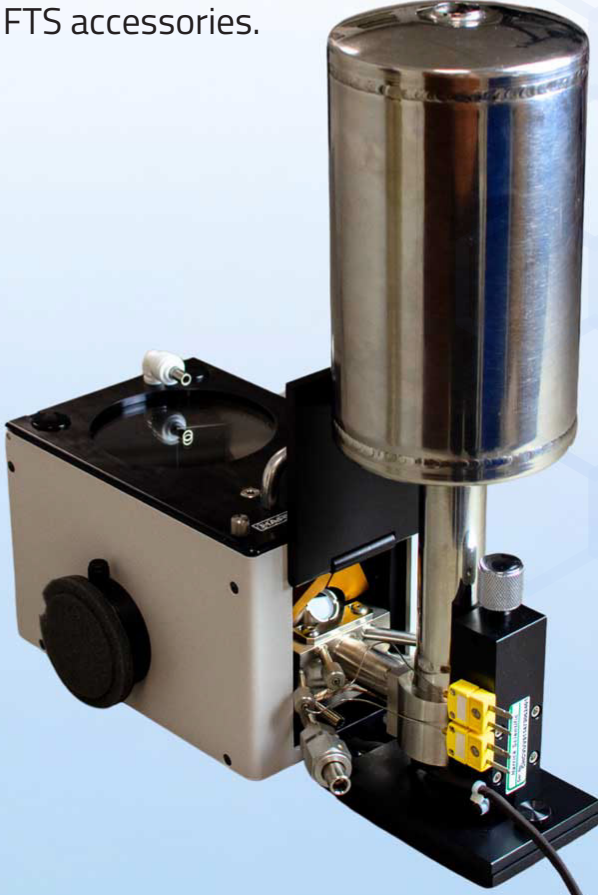
Unlock the full potential of your RWGS research with Specac's advanced DRIFTS accessories.

Performance excellence
Designed for accurate, real-time analysis of catalytic processes. 90° off-axis optical design efficiently collects diffusely reflecting light while minimising the specular reflection.

Robustness and versatility
Suitable for a wide range of applications, from academic research to industrial process optimisation. Handles extreme conditions (from -150°C to 910°C and up to 34 Bar of pressure).

Enhanced data quality
Provides reliable insights, enabling informed decision-making in catalyst development. Delivers pure data even for complex powdered catalytic samples.

Customisable components
Our team will happily tailor solutions to meet the unique needs of your project (e.g. designing custom low internal volume reaction chambers).



Ready to shift your catalyst research into new gear? Request a demo today.



Specac, a leading provider of spectroscopic equipment for advanced research and industrial applications, now offers the Harrick Scientific Praying Mantis™ accessory- designed to deliver high-quality data for demanding R&D environments.

References

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³ Global Synthetic Fuels Market to Soar to \$70.1 Billion by 2030, Driven by Rising Investments and Diversification in Primary Energy Mix, (accessed March 2025). <https://www.businesswire.com/news/home/20231030155725/en/Global-Synthetic-Fuels-Market-to-Soar-to-70.1-Billion-by-2030-Driven-by-Rising-Investments-and-Diversification-in-Primary-Energy-Mix---ResearchAndMarkets.com>

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